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HgWO₄ synthesized at high pressure and temperature

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Crystals of mercury(II) tungstate(VI), HgWO₄, were grown in sealed gold tubes under an Ar atmosphere at 300 MPa and 973 K. The monoclinic crystal structure (C2/c) of HgWO₄ consists of zigzag chains of edge-sharing WO₆ octahedra running along the *c* axis and layers of very distorted cornersharing HgO₆ octahedra in the *bc* plane. The Hg atom lies on an inversion centre and the W atom is on a twofold axis. No structural effects which can be ascribed to the high pressure used in the synthesis were found.

Comment

Tungstates are of interest for their luminescent properties, for use as detectors and calorimeters in high-energy experiments and in X-ray imaging. Strong absorption of high-energy radiation, in combination with a high density (9.2 Mg m⁻³), are some of the attractive properties of the title compound.

The crystal structure of HgWO₄ has recently been determined from neutron powder diffraction data by Rietveld refinements (Åsberg Dahlborg & Svensson, 2000). The two main differences between the previous powder study and the present single-crystal study are, first, that isotropic displacement parameters were used in the powder study, and secondly, the high pressure which was used for the growth of single crystals. In order to correlate structure with luminescent properties, precise structural information is needed, preferably from single-crystal data. This prompted us to perform the study which is presented herein.

HgWO₄ has a structure related to that of wolframite (Keeling, 1957) and is isostructural with HgMoO₄ (Jeitschko & Sleight, 1973). The structure consists of zigzag chains of edge-sharing WO₆ octahedra extending parallel to the *c* axis. There are two short W–O bonds [1.733 (5) Å], two medium [1.953 (4) Å] and two long bonds [2.197 (5) Å]. The O1–O1(-x, -y, -z) edge distance shared between connected WO₆ octahedra is only 2.372 (8) Å, while the other O–O distances are in the range 2.753 (8)–2.916 (8) Å. The Watom is displaced 0.337 Å from the centre of the WO₆ octahedron, calculated as the centre of gravity for the six O atoms.

inorganic compounds

In the *bc* plane, layers of corner-sharing HgO₆ octahedra are formed. The HgO₆ octahedra are very distorted, with two short collinear Hg–O bonds [2.044 (4) Å] and two pairs of long Hg–O distances [2.633 (5) and 2.743 (5) Å]. If the next O atoms, at a distance of 3.131 (5) Å from Hg, are included in the coordination, Hg is eight-coordinate. These HgO₈ polyhedra can be described as distorted cubes. The layer in the *bc* plane is then formed by edge-sharing HgO₈ cubes.

The only significant difference in bond distances in this study, compared with the neutron powder diffraction study, is the shorter W–W distance of 3.4143 (4) Å, compared with a distance of 3.425 (4) Å in the powder study. All the W–O distances are slightly shorter and all the Hg–O distances are





A polyhedral representation of $HgWO_4$ viewed along the *c* axis, with the *b* axis horizontal. The WO₆ octahedra are dark grey and the HgO_6 octahedra are light grey.



Figure 2

An ORTEP-3 plot (Farrugia, 1997) of the Hg and W coordination to O atoms in HgWO₄. Displacement ellipsoids are drawn at the 80% probability level [symmetry codes: (i) -x, -y, -z; (ii) -x, y, $\frac{1}{2} - z$; (iii) x, -y, $\frac{1}{2} + z$; (iv) x, y, 1 + z; (v) $\frac{1}{2} - x$, $\frac{1}{2} - y$, 1 - z; (vi) $\frac{1}{2} - x$, $\frac{1}{2} + y$, $\frac{3}{2} - z$; (vii) x, 1 - y, $\frac{1}{2} + z$; (viii) $\frac{1}{2} - x$, $\frac{1}{2} - y$, 2 - z; (ix) $\frac{1}{2} - x$, $y - \frac{1}{2}$, $\frac{3}{2} - z$].

slightly longer in the present single-crystal structure determination.

Displacement ellipsoids show that Hg has the largest displacements in the plane of the four long Hg–O bonds of the HgO₆ octahedron, as expected. U_{eq} is smaller for O1 than for Hg, which is in agreement with the neutron powder refinement. The displacement parameters for W are small and only slightly anisotropic.

Despite the differences in pressure, sample preparation and experimental data collection, there are very small differences between the results from the present single-crystal determination and the older structure determined from powder data. No effects which can be ascribed to the high pressure used for single-crystal synthesis have been found.

Experimental

Colourless prismatic crystals of HgWO₄ were synthesized under high gas pressure in collapsible gold tubes. The reaction mixture was composed of Na₂WO₄·2H₂O (Johnson Matthey, 95%) and HgCl₂ (Merck, 99.5%) in a 1:1 ratio. The reactants were loaded into a gold tube, which was then welded shut and placed inside a three-zone kanthal furnace, which was part of a high-pressure system (Lada *et al.*, 1998). The pressure was increased to 300 MPa using Ar as the pressure medium and the temperature was adjusted to 973 K. After 10 h, the temperature was decreased to room temperature over a period of a few hours, after which the pressure was decreased to 1 atm (1 atm = 101 325 Pa). The resulting prismatic crystals of HgWO₄ were washed with water. The largest crystals were about 1 mm in the longest direction.

Crystal data

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HgWO<sub>4</sub>

M_r = 448.44

Monoclinic, C2/c

a = 11.3791 (4) Å

b = 6.00794 (15) Å

c = 5.1456 (3) Å

\beta = 113.202 (3)°

V = 323.33 (2) Å<sup>3</sup>

Z = 4
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Data collection

Enraf–Nonius TurboCAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: Gaussian; a grid of 12 × 20 × 48 was used (Spek, 1990) $T_{min} = 0.078, T_{max} = 0.172$ 1750 measured reflections 713 independent reflections

Refinement

Refinement on F^2 R(F) = 0.027 $wR(F^2) = 0.070$ S = 1.10713 reflections 31 parameters $w = 1/[\sigma^2(F_o^2) + (0.0453P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $D_x = 9.212 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 25 reflections $\theta = 23.4-46.5^{\circ}$ $\mu = 82.80 \text{ mm}^{-1}$ T = 293 (2) K Prism, colourless $0.06 \times 0.03 \times 0.03 \text{ mm}$

620 reflections with $I > 2\sigma(I)$ $R_{int} = 0.053$ $\theta_{max} = 34.9^{\circ}$ $h = -18 \rightarrow 0$ $k = -9 \rightarrow 9$ $l = -7 \rightarrow 8$ 3 standard reflections frequency: 60 min intensity decay: none

 $\begin{array}{l} (\Delta/\sigma)_{\rm max} < 0.001 \\ \Delta\rho_{\rm max} = 3.10 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta\rho_{\rm min} = -3.13 \ {\rm e} \ {\rm \AA}^{-3} \\ {\rm Extinction \ correction: \ SHELXL97} \\ ({\rm Sheldrick, \ 1997}) \\ {\rm Extinction \ coefficient: \ 0.0031 \ (3)} \end{array}$

Table 1

Selected geometric parameters (Å, °).

W-O2	1.733 (5)	Hg-O2 ^{iv}	2.743 (5)
W-01	1.953 (4)	$Hg-O1^{v}$	3.131 (5)
W-O1 ⁱ	2.197 (5)	W-W ⁱ	3.4143 (4)
Hg-O1 ⁱⁱ	2.044 (4)	W-Hg	3.57426 (12)
Hg-O2 ⁱⁱⁱ	2.633 (5)	W-Hg ^{vi}	3.7082 (2)
01 W 01	(0.42 (10)		05 22 (10)
OI - W - OI	69.42 (19)	OI = Hg = O2	85.55 (18)
$OI^{-}W = OI^{-}W$	80.8 (2)	$OI-Hg-O2^{m}$	88.22 (17)
O1-W-O1 ^{vii}	84.61 (13)	O1 ⁱⁱ -Hg-O2 ⁱⁱⁱ	91.78 (17)
O2 ^{viii} -W-O1 ⁱ	89.3 (2)	O1-Hg-O2 ^{iv}	94.67 (18)
O2-W-O1	96.5 (2)	O2 ^{ix} -Hg-O2 ^{iv}	111.29 (14)
O2-W-O2viii	103.2 (4)	O1 ⁱⁱ -Hg-O1	180.0 (2)
O2 ^{viii} -W-O1	104.5 (2)	O2 ⁱⁱⁱ -Hg-O2 ^{ix}	180.0 (2)
O1-W-O1 ^{viii}	146.0 (3)	$O2^{iv} - Hg - O2^{x}$	180.00 (19)
O2-W-O1i	163.4 (2)	W-O2-Hg ^{xi}	119.9 (3)
O2 ⁱⁱⁱ -Hg-O2 ^{iv}	68.71 (14)	C .	

Symmetry codes: (i) -x, -y, -z; (ii) $\frac{1}{2} - x, \frac{1}{2} - y, -z$; (iii) $\frac{1}{2} - x, \frac{1}{2} - y, 1 - z$; (iv) $\frac{1}{2} - x, y - \frac{1}{2}, \frac{1}{2} - z$; (v) $x, -y, z - \frac{1}{2}$; (vi) $x - \frac{1}{2}, y - \frac{1}{2}, z$; (vii) $x, -y, \frac{1}{2} + z$; (viii) $-x, y, \frac{1}{2} - z$; (ix) x, y, z - 1; (x) $x, 1 - y, z - \frac{1}{2}$; (xi) x, y, 1 + z.

The maximum extinction correction (*SHELXL97*; Sheldrick, 1997) was y = 0.84 for the 111 reflection (the observed structure factor is $F_{\rm obs} = yF_{\rm kin}$, where $F_{\rm kin}$ is the kinematic value of the structure factor). The highest residual electron-density peak was located 0.95 Å from O2 and 0.97 Å from W. There were also peaks between Hg and O1.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *LATCON* (Schwarzenbach & King, 1999); data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97; molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: IZ1018). Services for accessing these data are described at the back of the journal.

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